R16 Code No. 132AG JAWAHARLAL NEHRÜ TECHNOLOĞICAL UNIVERSITY HYDERABAD B. Tech I Year II Semester Examinations, August/September - 2017 **ENGINEERING CHEMISTRY** (Common to CE, ME, MCT, MMT, MIE, CEE, MSNT) Time: 3 hours Max. Marks: 75 Note: This guestion paper contains two parts A and B. Part A is compulsory which carries 25 marks. Answer all questions in Part Part B consists of 5 Units. Answer any one full question from each unit. Each question carries 10 marks and may have a, b, c as sub questions. PART- A (25 Marks) Write the specifications of potable water. What is Caustic embrittlement? Give chemical reaction involved [3] c) What are secondary cells? Give two examples. [2] What is single electrode potential? Write Nernst equation to calculate electrode potential. d) [3] Give the characteristics of Elastomers. e) [2] Write short note on free radical addition polymerisation. f) [3] Explain the importance of Octane number. [2] g) What is CNG? Give its composition and characteristic properties. i) Define flash and fire point of a lubricant. [2] j) What are special cements? Give their uses. [3] **PART-B** Explain the Break point chlorination and give its significance. A Sample of water showed the following analysis $CaCl_2 = 2.22mg/l$; $Mg(NO_3)_2 = 1.48mg/l;$ $KCl = 7.45 mg/l; CaSO_4 = 1.36 mg/l;$ $Mg(HCO_3) = 2.92 \text{ mg/l}$; Organic impurities = 1.22mg/l. Calculate the temporary and permanent hardness of the given water sample. [5+5]OR Differentiate between chlorination and ozonization disinfection methods of potable water. b) Explain the steps involved in treatment of sewage water. [5+5] 4.a) What are fuel cells? Explain the construction and working of hydrogen oxygen fuel cell. What do you understand by electrochemical series? Explain its applications. b) OR Explain the construction, working and application of glass electrode with neat diagram. Explain the chemical reactions of lead acid battery during its charging and discharging. [5+5]

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